## X-Ray Crystal Structure and Configuration of Enhydrin Bromohydrin

By GOPINATH KARTHA and K. T. GO\*

(Center for Crystallographic Research, Roswell Park Memorial Institute, Buffalo, New York 14203)

and B. S. JOSHI

(CIBA Research Centre Goregaon, Bombay 63, India)

Summary An X-ray study has shown that  $C_{23}H_{29}O_{10}Br$  is a germacronolide sesquiterpene lactone having a *cis* stereochemistry at the double bond C-1–C-10, and an epoxide ring at positions C-4,C-5 which has a *trans* configuration; enhydrin thus falls into group 2 of the classification recently proposed by Neidle and Rogers.<sup>1</sup>

The sesquiterpene lactone named enhydrin was first isolated from *Enhydra fluctuans*<sup>2</sup> a herb which has been used for the treatment of skin diseases and nervous ailments, and also as a laxative.<sup>3</sup> The gross structure (I)<sup>†</sup> of enhydrin was reported<sup>4</sup> on the basis of degradative and spectroscopic evidence.<sup>†</sup>



An X-ray study was undertaken to determine the detailed configuration of the enhydrin molecule and to make use of the assignment of groups at C-8 and C-9. X-ray crystallographic studies were made on the bromohydrin obtained by reaction of enhydrin with 40% HBr in methanol, which reaction opened the epoxide ring of the side chain.

Crystals of enhydrin bromohydrin  $C_{23}H_{29}O_{10}Br$  are orthorhombic and have a space group  $P2_12_12_1$ , a = 10.08, b = 26.97, c = 8.94 Å and Z = 4. Intensity data were collected manually on a diffractometer with  $Cu-K_{\alpha}$  radiation to a Bragg angle of 75°, using the stationary-crystal stationary-counter method and balanced Ni and Co filter pairs. Of the 2957 reflections measured, 2474 were considered observed. The structure was solved by direct methods<sup>6,7</sup> and refined by block-diagonal least-squares to an R value of 0.148, the bromine atom alone being ascribed anisotropic thermal parameters.

A view of the molecule and the stereochemistry of bromo-



Table.	Torsional	angles about	the bonds in the ten-mem	bered ring
C-1-C-2-	-C-3–C-4	$+75^{\circ}$	C-6-C-7-C-8-C-9	$-63^{\circ}$
C-2-C-3-	-C-4–C-5	-92°	C-7-C-8-C-9-C-10	$-56^{\circ}$
C-3C-4-	-C-5-C-6	$+147^{\circ}$	C-8-C-9-C-10-C-1	$+132^{\circ}$
C-4-C-5-	-C-6C-7	$-119^{\circ}$	C-9-C-10-C-1-C-2	+13°
C-5-C-6-	-C-7–C-8	$+95^{\circ}$	C-10-C-1-C-2-C-3	-97°



FIGURE. The bromohydrin enhydrin molecule viewed approximately normal to the macrocycle.

<sup>†</sup> Drawn according to the convention of Rogers et al.<sup>5</sup>

enhydrin (II)<sup>†</sup> is shown in the Figure. The general features agree with those suggested by Joshi et al.4 and confirm the assignment of the acetyl and  $\alpha$ -methyl- $\alpha,\beta$ epoxy-butyl group group at positions C-8 and C-9 of the ring respectively. The occupancy of cis and trans configurations within the ring make the ten-membered macrocycle very distorted. The torsion angles<sup>8</sup> about the bonds in this ring (Table) show that the epoxide ring at position C-4,C-5 is trans fused whereas the double bond C-1-C-10 has a cis stereochemistry. Thus enhydrin falls in the group 2 classification of Neidle<sup>1</sup> et al., and in this respect resembles melampodin, the only germacronolide with this pattern of endocyclic double bond previously studied by X-ray diffraction. As in melampodin whose stereochemistry it resembles closely, it is seen that C-14 and C-16 are anti with respect to the plane of macrocycle. The relative configurations about the asymmetric atoms are 4R, 5R, 6S, 7S, 8S, 9S, 20S, and 22S. We plan to determine the absolute configuration using anomalous scattering from the bromine atom.

We thank Mrs. Consuelo Vincent for help in collecting the diffraction data and the New York State Department of Health, the National Science Foundation and National Institute of Health for their support.

(Received, 21st August 1972; Com. 1459.)

- <sup>1</sup> S. Neidle and D. Rogers, J.C.S. Chem. Comm., 1972, 140.
  <sup>2</sup> N. R. Krishnaswamy, T. R. Sheshadri, and B. R. Sharma, Current Sci., 1968, 37, 94.
  <sup>3</sup> R. N. Chopra, S. L. Nayar, and I. C. Chopra, 'Glossary of Indian Medicinal Plants,' C.S.I.R., New Delhi, 1956, 107.
  <sup>4</sup> B. S. Joshi, V. N. Kamat, and H. Fuhrer, Tetrahedron Letters, 1971, 26, 2373.
  <sup>5</sup> D. Rogers, G. P. Moss, and S. Neidle, J.C.S. Chem. Comm., 1972, 142.
  <sup>6</sup> J. Karle and I. L. Karle, Acta Cryst., 1966, 21, 849.
  <sup>7</sup> H. Hauptman and J. Karle, "Solution of the Phase Problem," A.C.A. Monograph 3, Polycrystal Book Service.
  <sup>8</sup> N. Klupa and V. Prelog, Exterinita 1960, 16, 521.

- <sup>8</sup> N. Klyne and V. Prelog, Experientia, 1960, 16, 521.